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Application of GDOES to Study of Corrosion Protective Coatings

Ruth Bingham, George Thompson, Peter Skeldon.

The University of Manchester

GD day 16th September 2016





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Project background

MULTIPROTECT:

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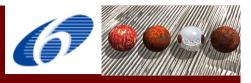
• An integrated project of the 6th framework programme of the European Commission. • A collaboration between 31 academic and industrial researchers from 13 countries. • Aim: to provide novel, heavy-metal free, multifunctional, economical and environmentally friendly corrosion protecting surface treatments based on nanocomposite materials. http://www.multiprotect.org/



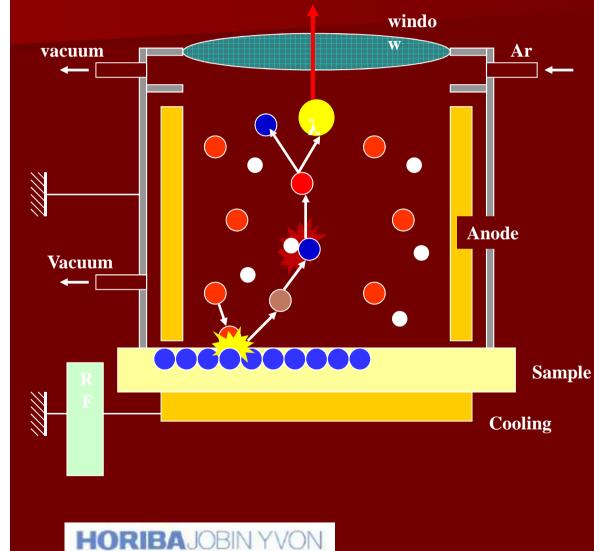
Specimens: AA2024 T3

- EPOXY-Al sol-gel coatings (INM)
 - Non-inhibited

- Inhibited: BZT or BZT in AluOx;
- Methacryloxy-based sol-gel coatings (EADS)
 - Non-inhibited
 - SAPP-inhibitor in sol-gel,
 - Inhibitors in primer: BZT, MBT, MBI, SrCrO₄.



Experimental: RF-GDOES



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- The sample is sputtered by Ar ions.
- Neutral species are accelerated in the plasma.
- Light is emitted by the atoms when excited by collisions.



Experimental: Artificial Cells

• Cells consisted of coated specimens fixed in close proximity to the bare alloy:

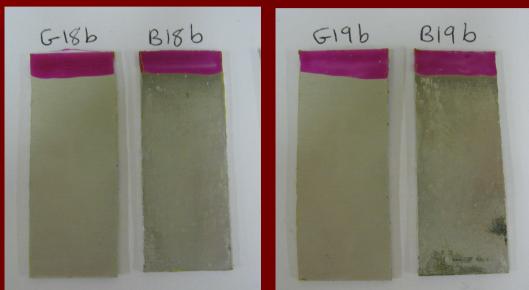


Cells were filled with 35 g l⁻¹ NaCl solution for (a) 5 h and (b) 200 h.
Subsequent GDOES analysis of the bare alloy from each cell.
Noted characteristics of elemental profiles.

Experimental: Artificial Cells



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Artificial cellsduring exposure

^ Artificial cells deconstructed after exposure

Bare alloy specimens were then sputtered. Raw data was imported to Excel and combined to compare profiles of individual elements.

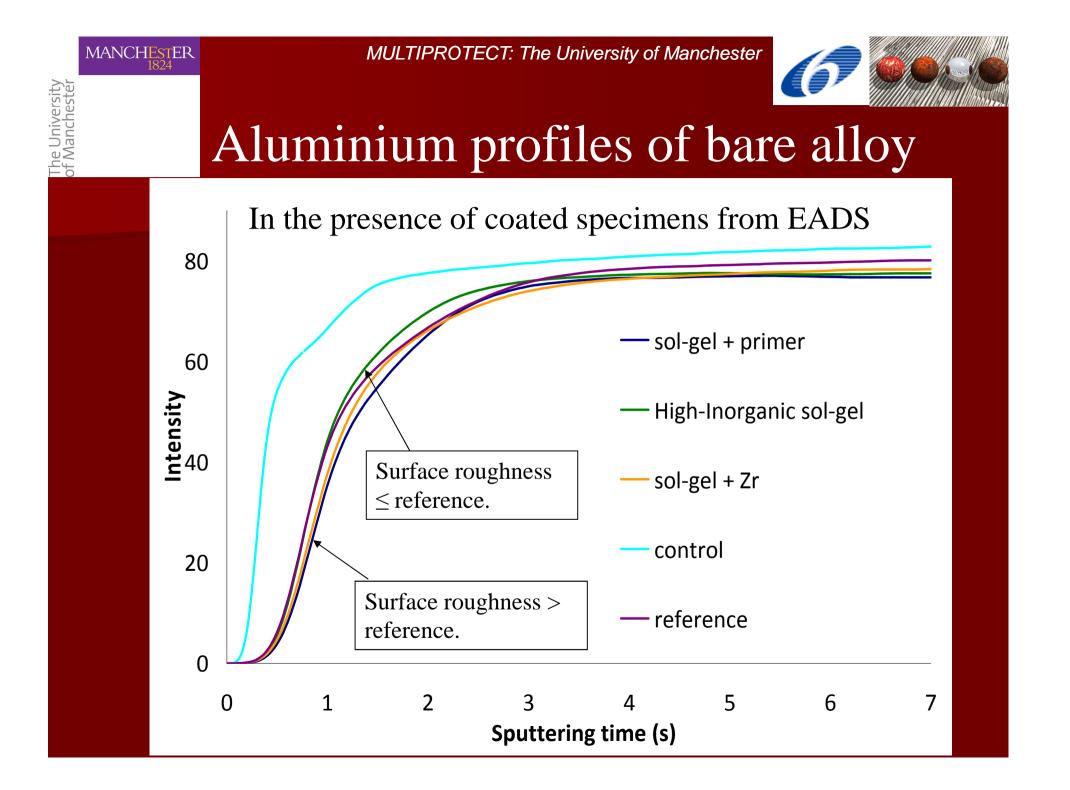


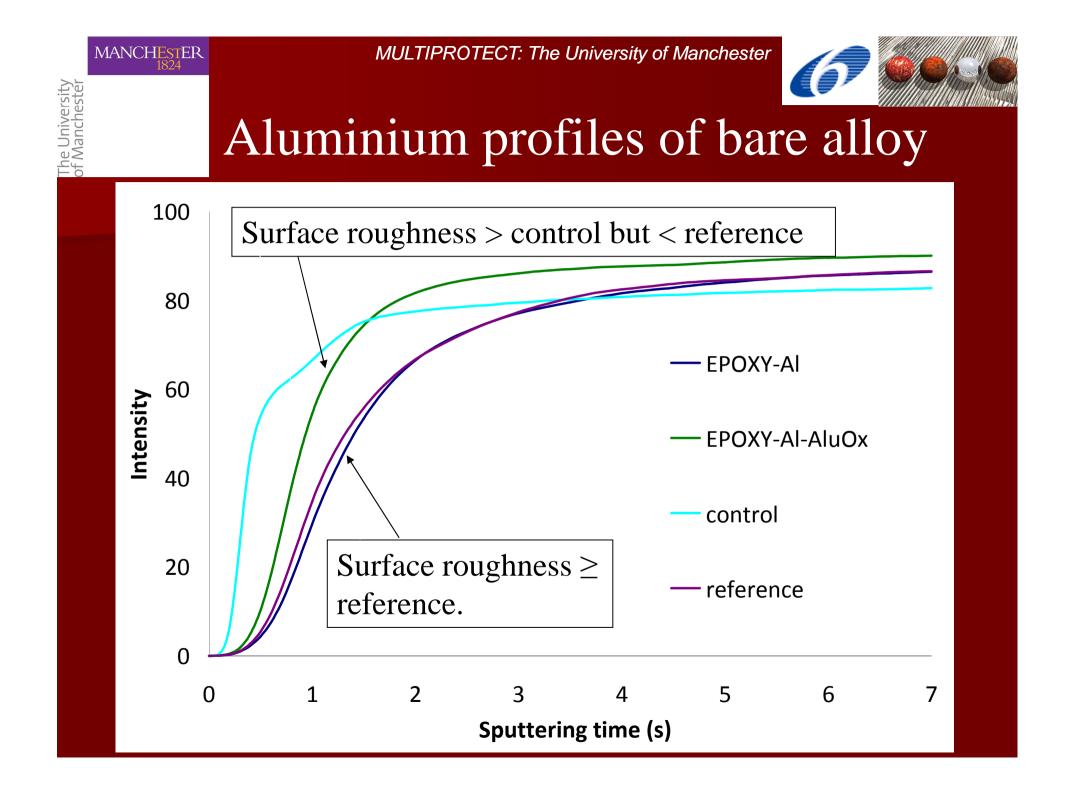


Results and Discussion

Part 1: Elemental depth profiles of the bare alloy from cells incorporating uninhibited coatings

Surf. Interface Anal. **2015**, 47, 1009 – 1014

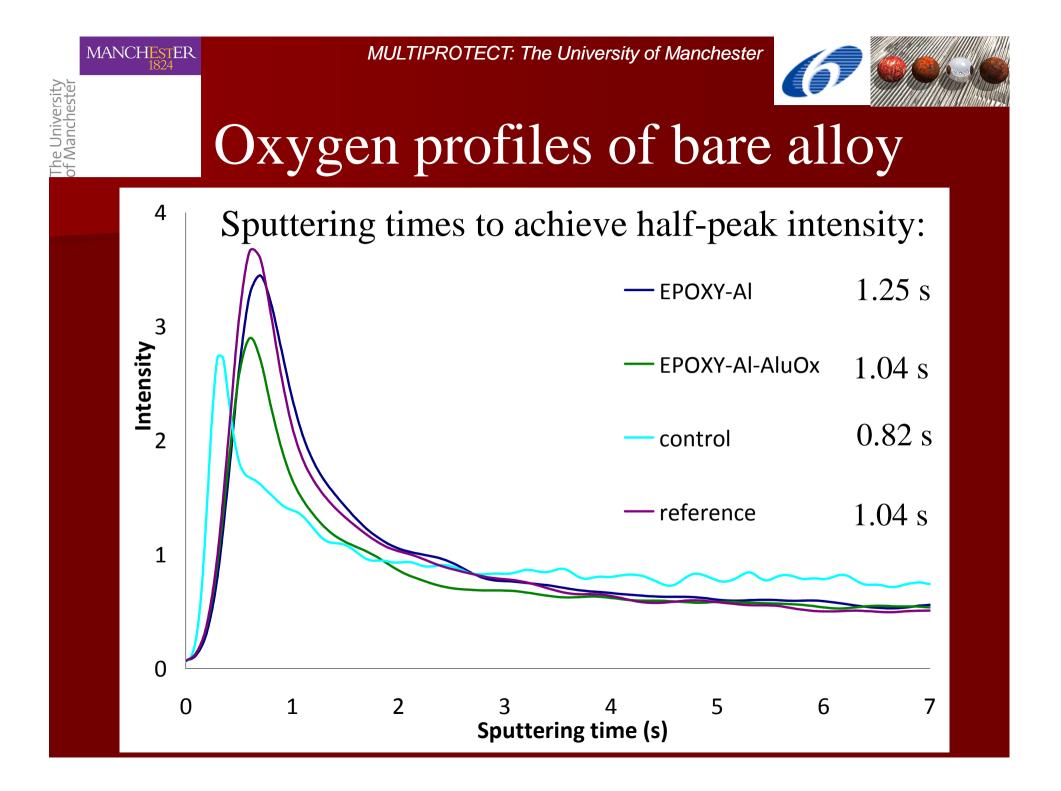






Oxygen profiles of bare alloy

8	Sputtering times to achieve half-peak intensity:						
		Area und		_	— sol-gel + prime	er]	l.15 s
htensity b		curve indicates thickness of corroded or altered layer		- d _	 High-Inorganic so sol-gel + Zr 		^{gel} 1.15 s 1.06 s
		layer		-	- control]	1.04 s
2					– reference		1.24 s
0	J						
() 1	2	3	4	5	6	7
Sputtering time (s)							

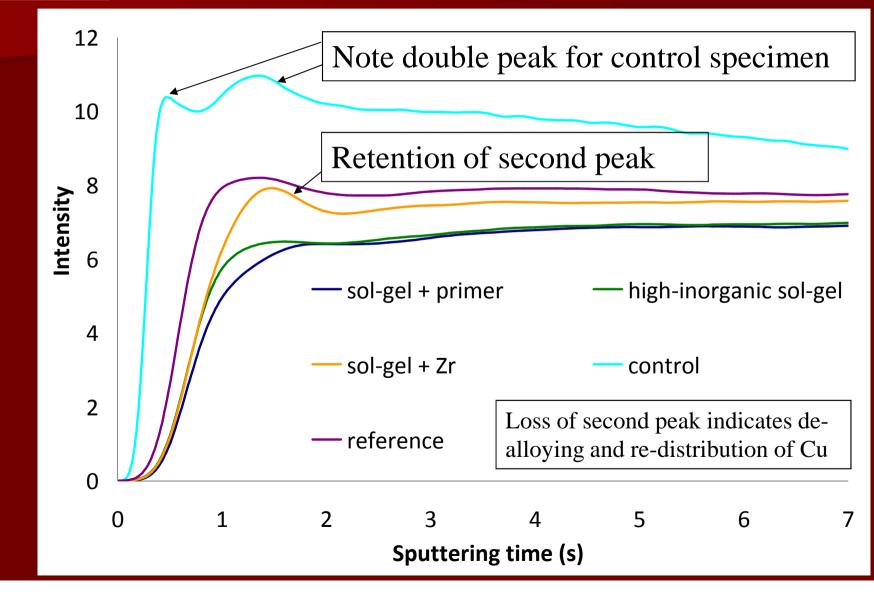




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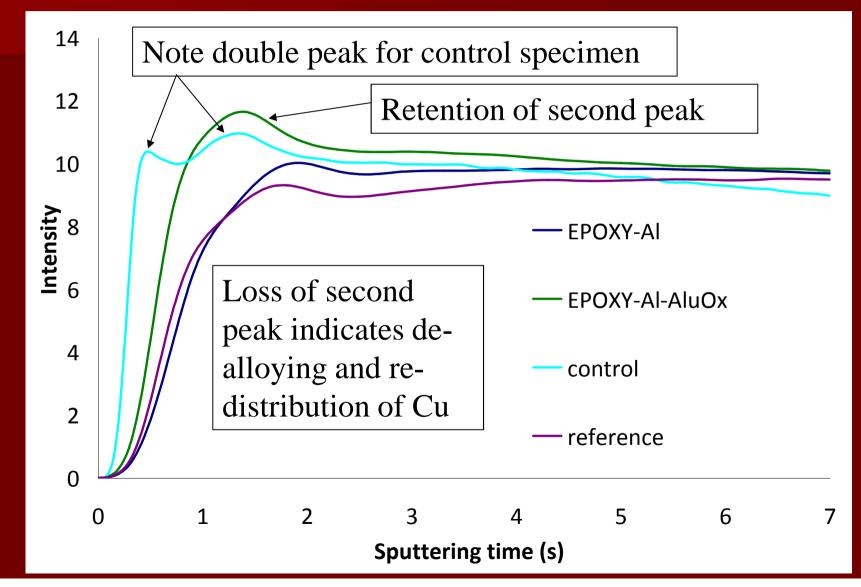


Copper profiles of bare alloy





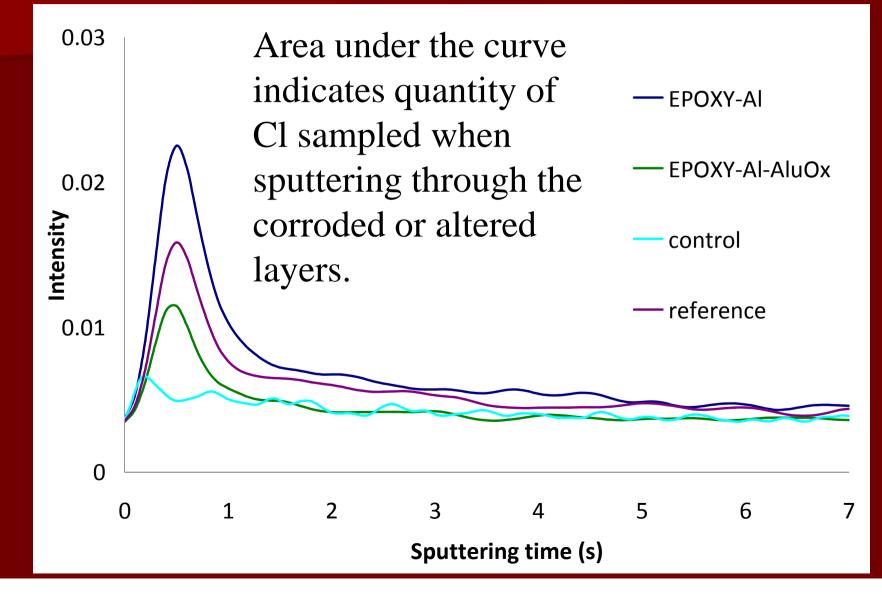
Copper profiles of bare alloy



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Chlorine profiles of bare alloy







Results and Discussion

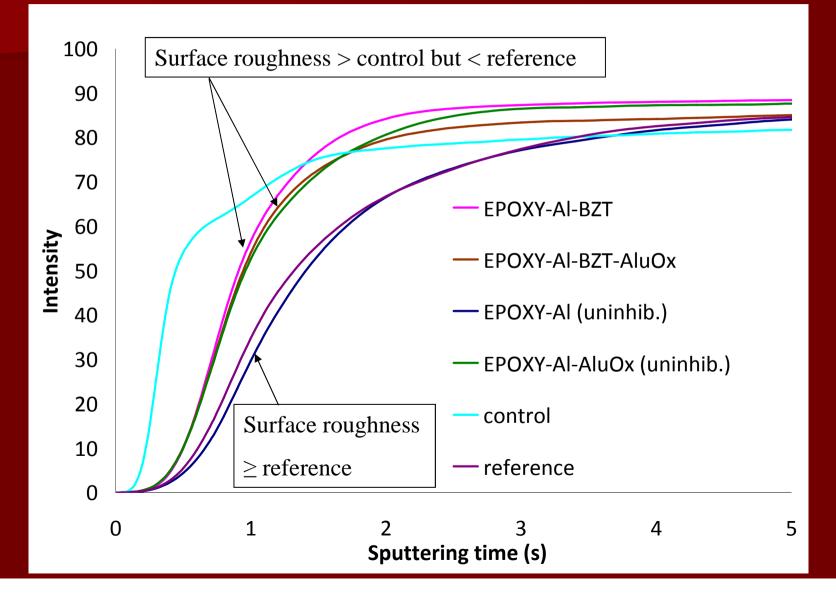
Part 2: Elemental depth profiles of the bare alloy from cells incorporating BZTinhibited coatings (INM)

Surf. Interface Anal. **2015**, 47, 1098 – 1108

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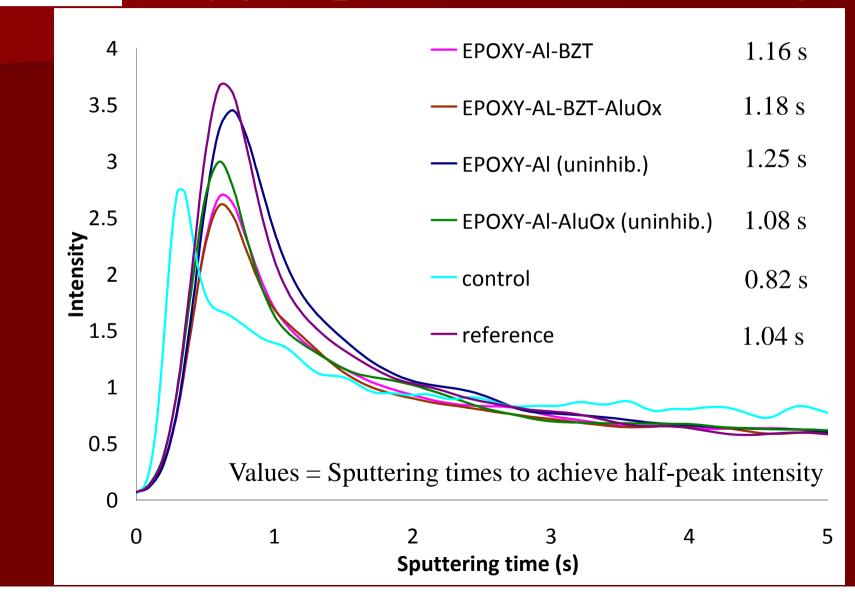


Aluminium profiles of bare alloy





Oxygen profiles of bare alloy

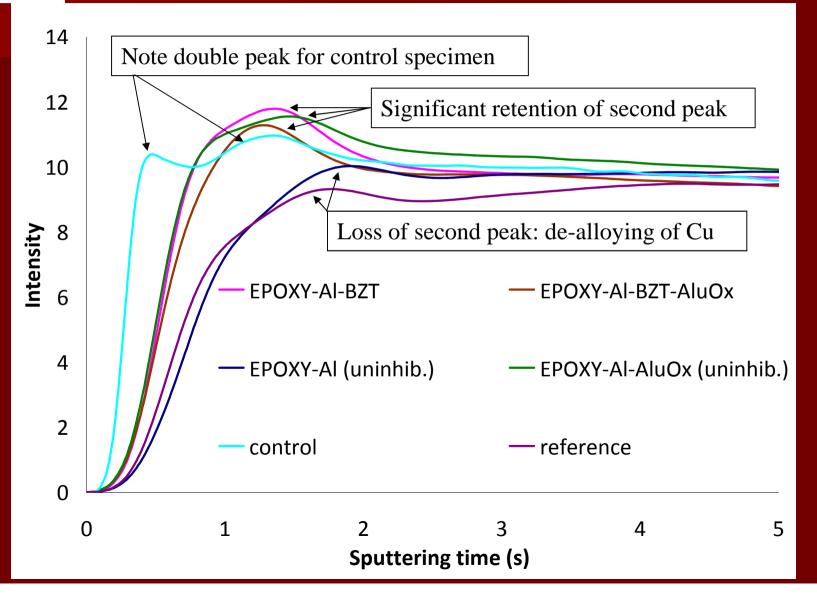




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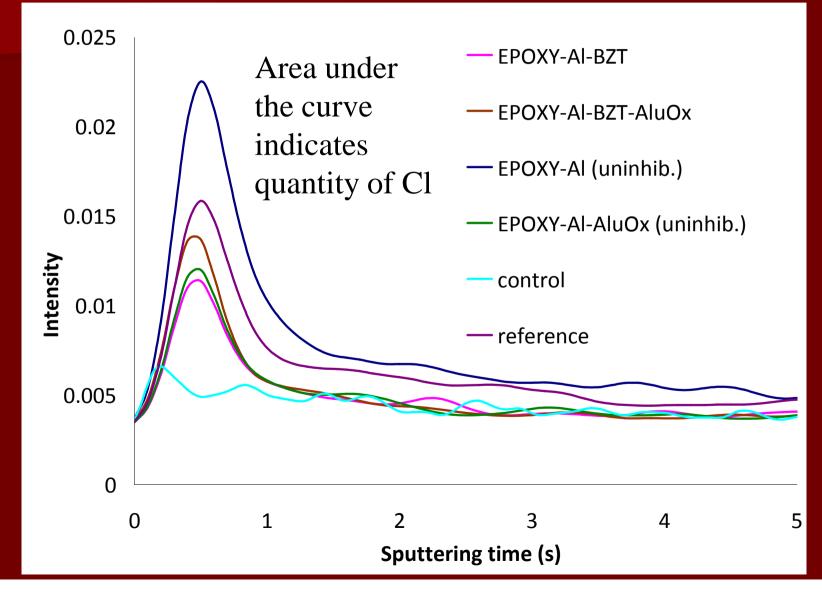
Copper profiles of bare alloy



MULTIPROTECT: The University of Manchester



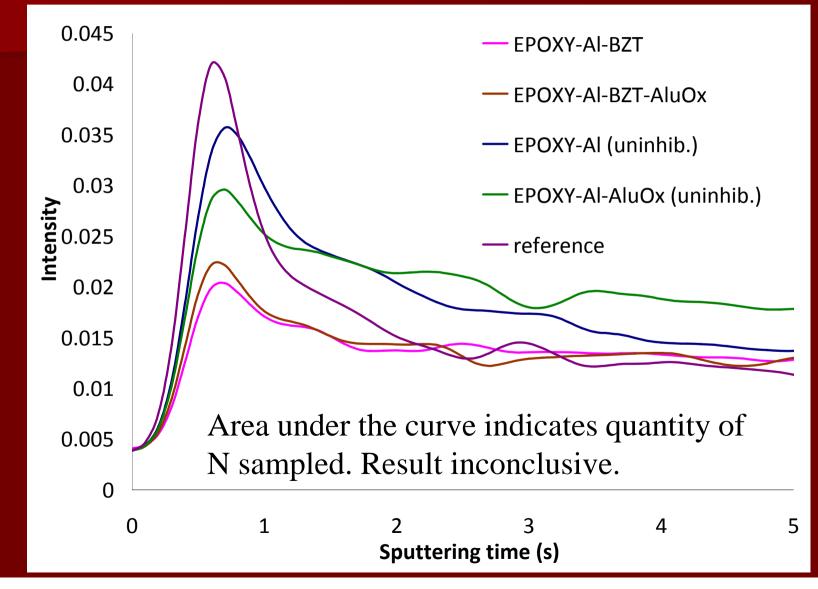
Chlorine profiles of bare alloy

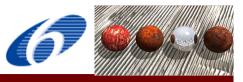


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Nitrogen profiles of bare alloy





Summary and conclusions

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Comparison of the intensity of corrosion of the bare alloy can be made by inspection of Al depth profiles.
Comparison of thickness of corroded layers can be made by inspection of the O depth profiles.
Assessment of extent of de-alloying and re-distribution of Cu can be made by inspection of depth profiles.
An inhibitory effect of the empty AluOx nanocontainers, due to chloride absorption, was inferred rom the Cl depth profiles.

• Presence of BZT inhibitor could not be confirmed by inspection of N depth profiles.



Additional Work

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A third paper has been published detailing a study of depth profiles obtained by sputtering the bare alloy after corrosion in the presence of the inhibitor-doped specimens from EADS:

Surf. Interface Anal. **2016**, 48, 341 – 352





Thank you...

...any questions?